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#### **Key indicators**

Single-crystal X-ray study T = 130 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.019 wR factor = 0.046 Data-to-parameter ratio = 21.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# Dichlorobis(pyridinium-2-thiolato)cadmium(II)

The title compound,  $[Cd(C_5H_5NS)_2Cl_2]$ , was prepared by the self-assembly reaction of isophthalic acid,  $CdCl_2 \cdot 5H_2O$  and pyridine-2-thiol in methanol/H<sub>2</sub>O. The Cd<sup>II</sup> atom lies on a twofold axis and is tetrahedrally coordinated by two S atoms from two pyridinium-2-thiolate ligands and two Cl atoms. Molecules are linked by N-H···Cl hydrogen bonds to form sheets parallel to the *bc* plane.

#### Comment

The coordination chemistry of the pyridine-2-thiol ligand with transition metals has undergone extensive development in recent years owing to its abundant coordination ability. The pyridine-2-thiol ligand can coordinate both in the 'thiolate' and the tautomeric 'thione' forms and it has several binding modes, for instance, monodentate, bidentate chelating, bidentate bridging and doubly bridging. Recently, some coordination polymers constructed from pyridine-2-thione (or pyridinium-2-thiolate) have been reported (Lobana *et al.*, 2000; Hong *et al.*, 1999; Berardini *et al.*, 1997). We report here the synthesis and crystal structure of the title mononuclear cadmium(II) compound, (I).



As shown in Fig. 1, the Cd atom in (I) lies on a twofold axis and is coordinated by two S atoms from two pyridinium-2thiolate ligands and two Cl atoms in a distorted tetrahedral geometry, with Cd–S and Cd–Cl distances of 2.5172 (9) and 2.4730 (9) Å, respectively. The title compound is isostructural with the Co analog,  $[Co(C_5H_5NS)_2Cl_2]$  (Binamira-Soriaga *et al.*, 1979).



**Figure 1** A view of the structure of (I), showing 50% probability displacement ellipsoids [symmetry code: (i) -x, y,  $\frac{1}{2} - z$ ].

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Figure 2

Packing diagram of (I), showing the two-dimensional hydrogen-bonded structure. H atoms not involved in hydrogen bonding have been omitted and the hydrogen bonds are shown as dashed lines.

In the crystal structure, there are  $N-H\cdots Cl$  hydrogen bonds involving the protonated N atoms and the Cl atoms from symmetry-related molecules (see Table 2). As shown in Fig. 2, molecules are linked by these interactions to form twodimensional sheets parallel to the *bc* plane.

## **Experimental**

The title compound was prepared by mixing a 1:1 molar ratio of isophthalic acid (0.083 g, 0.5 mmol) and CdCl<sub>2</sub>·5H<sub>2</sub>O (0.114 g, 0.5 mmol) in a solution of MeOH/H<sub>2</sub>O ( $3:2 \nu/\nu$ , 25 ml). The solution was stirred at 333 K for 30 min, pyridine-2-thiol (0.056 g, 0.5 mmol) was added and the resulting mixture stirred for a further 1 h. The reaction mixture was filtered and colorless single crystals suitable for X-ray analysis were obtained by slow evaporation of the solvent.

#### Crystal data

$[Cd(C_5H_5NS)_2Cl_2]$	$D_x = 1.949 \text{ Mg m}^{-3}$
$M_r = 405.62$	Mo $K\alpha$ radiation
Monoclinic, $C2/c$	Cell parameters from 2053
a = 9.364 (4)  Å	reflections
b = 11.445 (4)  Å	$\theta = 2.9 - 28.3^{\circ}$
c = 13.597 (6) Å	$\mu = 2.25 \text{ mm}^{-1}$
$\beta = 108.46 \ (2)^{\circ}$	T = 130 (2)  K
$V = 1382.1 (10) \text{ Å}^3$	Prism, colorless
Z = 4	$0.50 \times 0.25 \times 0.08 \text{ mm}$
Data collection	
Bruker SMART CCD	1709 independent reflections
diffractometer	1659 reflections with $I > 2\sigma(I)$
$\omega$ scans	$R_{\rm int} = 0.020$
Absorption correction: multi-scan	$\theta_{\rm max} = 28.3^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -12 \rightarrow 6$

 $\begin{array}{l} k=-15\rightarrow 15\\ l=-16\rightarrow 18 \end{array}$ 

$T_{\min} = 0.515, T_{\max} = 0.836$
5486 measured reflections

Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0221P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.019$	+ 1.6413P]
$vR(F^2) = 0.046$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.09	$(\Delta/\sigma)_{\rm max} = 0.002$
709 reflections	$\Delta \rho_{\rm max} = 0.67 \ {\rm e} \ {\rm \AA}^{-3}$
78 parameters	$\Delta \rho_{\rm min} = -0.65 \text{ e } \text{\AA}^{-3}$

# Table 1 Selected geometric parameters (Å, °).

Cd1-Cl1	2.4730 (9)	Cd1-S1	2.5172 (9)
Cl1 <sup>i</sup> -Cd1-Cl1	110.87 (4)	Cl1-Cd1-S1	108.11 (3)
Cl1 <sup>i</sup> -Cd1-S1	112.83 (2)	S1-Cd1-S1 <sup>i</sup>	103.97 (4)

Symmetry code: (i) -x, y,  $\frac{1}{2} - z$ .

Table 2Hydrogen-bonding geometry (Å,  $^{\circ}$ ).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
N1-H1B···Cl1 <sup>ii</sup>	0.86	2.30	3.1487 (18)	171
Symmetry codes: (ii) 1	$\pm r^{1} \pm n^{2}$			

Symmetry codes: (ii)  $\frac{1}{2} + x, \frac{1}{2} + y, z$ .

All H atoms were included in calculated positions and refined using the riding-model approximation  $[C-H = 0.93 \text{ Å}, N-H = 0.86 \text{ Å} \text{ and } U_{iso}(H) = 1.2U_{eq}(C,N)].$ 

Data collection: *SMART* (Siemens, 1996); cell refinement: *SMART* and *SAINT* (Siemens, 1994); data reduction: *SAINT* and *XPREP* in *SHELXTL* (Siemens, 1994); program(s) used to solve structure: *SHELXTL*; program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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### References

Berardini, M., Lee, J., Freedman, D., Lee, J., Emge, T. J. & Brennan, J. G. (1997). *Inorg. Chem.* **36**, 5772–5776.

Binamira-Soriaga, E., Lundeen, M. & Seff, K. (1979). Acta Cryst. B35, 2875–2879.

Hong, M., Su, W., Cao, R., Zhang, W. & Lu, J. (1999). *Inorg. Chem.* **38**, 600–602.

Lobana, T. S., Verma, R., Hundal, G. & Castineiras, A. (2000). *Polyhedron*, **19**, 899–906.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Siemens (1994). SAINT and SHELXTL. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Siemens (1996). SMART. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.